

## Product Name MACHINE DISHWASH CHLORINATED LIQUID

## 1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

Supplier Name CLEAN PLUS CHEMICALS PTY LTD

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Synonym(s) AUTO DISHWASHING CHLORINATED LIQUID • PRODUCT CODE – 130

Use(s) AUTOMATIC DISHWASHING DETERGENT.SANITISER.

**SDS Date** 4 May 2010 v2

26 June 2012 v3

#### 2. HAZARDS IDENTIFICATION

#### CLASSIFIED AS HAZARDOUS ACCORDING TO NOHSC/ASCC CRITERIA

**RISK PHRASES** 

R31 Contact with acids liberates toxic gas

R35 Causes severe burns

R41 Risk of serious damage to eyes

**SAFETY PHRASES** 

S1/2 Keep locked up and out of reach of children

S26 In case of contact with eyes, rinse immediately with plenty of water and seek medical advice

S28 After contact with skin wash immediately with plenty of soap suds

S37/39 Wear suitable gloves and eye/face protection

S45 In case of accident or if you feel unwell seek medical advice immediately (show the label where possible)

S50 Do not mix with incompatible materials

CLASSIFIED AS A DANGEROUS GOODS BY THE CRITERIA OF THE ADG CODE

UN No. 1760 DG Class 8 Subsidiary Risk(s) None Allocated

Packing Group III Hazchem Code 2R EPG 8A1

## 3. COMPOSITION/INFORMATION ON INGREDIENTS



SODIUM HYDROXIDE	Na-OH	1310-73-2	10-30%
SODIUM HYPOCHLORITE	CI-O.Na	7681-52-9	10-30%
NON HAZARDOUS INGREDIENTS	Not Available	Not Available	Remainder

## 4. FIRST AID MEASURES

Eye If in eyes, hold eyelids apart and flush the eye continuously with running water. Continue flushing until advised to

stop by the Poison Information Centre or a doctor, or for at least 15 minutes.

**Skin** If skin or hair contact occurs, remove contaminated clothing and flush skin and hair with running water. Continue

flushing with water until advised to stop by the Poisons Information Centre or a doctor.

Inhalation If inhaled, remove from contaminated area. To protect rescuer, use an Air-line respirator where an inhalation risk

exists. Apply artificial respiration if not breathing.

Ingestion For advice, contact a Poison Information Centre on 13 11 26 (Australia Wide) or a doctor (at once). If swallowed, do

not induce vomiting.

Advice to Doctor CORROSIVE POISONING TREATMENT: Immediate treatment preferably in a hospital is mandatory. In treating

corrosive poisoning. DO NOT INDUCE VOMITING; DO NOT ATTEMPT GASTRIC LAVAGE; and DO NOT ATTEMPT TO NEUTRALISE THE CORROSIVE SUBSTANCE. Vomiting will increase the severity of damage to the oespohagus as the corrosive substance will again come in contact with it. Attempting gastric lavage may result

in perforating either the oesophagus or stomach.

First Aid Facilities Eye wash facilities and safety shower should be available.

#### 5. FIRE FIGHTING MEASURES

Flammability Non flammable. May evolve toxic gases if strongly heated to decomposition. Contact with some metals (eg:

aluminum), may liberate potentially flammable – explosive hydrogen gas.

Fire and Explosion Non flammable. May evolve flammable hydrogen gas in contact with some metals. If product is present in a

fire, toxic gases may be evolved. Evacuate area and contact emergency services. Remain upwind & notify those downwind of hazard. Wear full protective equipment including Self Contained Breathing Apparatus

(SCBA) when combating fire. Use waterfog to cool intact containers and nearby storage areas.

**Extinguishing** Non flammable. Prevent contamination of drains or waterways.

Hazchem Code 2R

## 6. ACCIDENTAL RELEASE MEASURES

Spillage

If spilt (bulk), contact emergency services if appropriate. Wear splash-proof goggles and PVC/rubber gloves, an Airline respirator (where an inhalation risk exists). Absorb spill with sand or similar and place in sealed containers for disposal. Wash spill site down with water. For small amounts, dilute with water and flush to sewer. Caution: surfaces may be slippery.

### 7. STORAGE AND HANDLING

Storage

Store in cool, dry, well ventilated area, removed from acids, combustible materials and foodstuffs. Ensure containers are adequately labeled, protected from physical damage and sealed when not in use. Check regularly for leaks or spills. Large storage areas should be bunded and have appropriate ventilation systems.

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**Handling** Before use carefully read the product label. Use of safe work practices are recommended to avoid eye or skin contact

and inhalation. Observe good personal hygiene, including washing hands before eating. Prohibit eating, drinking and

smoking in contaminated areas.

## 8. EXPOSURE CONTROLS/ PERSONAL PROTECTION

Exposure Stds Sodium Hydroxide (Peak limitation) TWA: 2.0mg/m<sup>3</sup> (Reference: ASCC(AUS))

Sodium Hypochlorite TWA: 1.0 ppm 3.0mg/m³ (Reference: ASCC(AUS))

Biological Limits No biological limit allocated.

Engineering Controls Ensure adequate natural ventilation. Maintain vapour levels below the recommended exposure standard.

PPE Wear splash-proof goggles and PVC or rubber gloves, rubber, face shield and coveralls. Where an

inhalation risk exists, wear a type A (organic vapor) respirator.

### 9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance CLEAR LIQUID Solubility (Water) SOLUBLE

OdourSLIGHT ODOURSpecific Gravity1.24 – 1.26

Ph 13.5 – 14.0 Volatiles NOT AVAILABLE

Vapour PressureNOT AVAILABLEFlammabilityNON FLAMMABLE

 Vapour Density
 NOT AVAILABLE
 Flash Point
 NOT RELEVANT

**Boiling Point** 100°C (Approximately) **Upper Explosion Limit** NOT RELEVANT

 Melting Point
 NOT AVAILABLE
 Lower Explosion Limit
 NOT RELEVANT

Evaporation Rate NOT AVAILABLE

#### 10. STABILITY AND REACTIVITY

Chemical Stability Stable under recommended conditions of storage.

**Conditions to Avoid** Avoid heat, sparks, open flames and other ignition sources.

Material to Avoid Incompatible with oxidizing agent (e.g. hypochlorite, peroxide), acids (eg. Hydrochloric acid), heat

and ignition sources. Also incompatible with combustible materials and dangerous goods.

**Decomposition** May evolve toxic gas if heated to decomposition.

**Hazardous Reactions** Polymerization is not expected to occur.

#### 11. TOXICOLOGICAL INFORMATION

**Health Hazard** Highly corrosive. Use safe work practices to avoid eye or skin contact, spray mist generation or inhalation.

**Eye** Highly corrosive – Severe irritant. Contact may result in irritation, lacrimation, pain, redness and conjunctivitis.

Prolonged contact may result in corneal burns and possible permanent damage.



**Inhalation** Corrosive - severe irritant. Over exposure to mists or vapours (if sprayed) may result in mucous membrane

irritation of the nose and throat with coughing. At high levels nausea, dizziness and headache. Low vapour

pressure, considerably reduces the potential for an inhalation hazard.

**Skin** Corrosive. Prolonged or repeated contact may result in drying the skin, rash and dermatitis.

Ingestion Highly corrosive. Ingestion may result in burns to the mouth and throat, nausea, vomiting, abdominal pain and

diarrhea. Ingestion of large quantities may result in ulceration, unconsciousness, convulsion, and death.

Toxicity Data SODIUM HYDROXIDE (1310-73-2)

LDLo (Ingestion):500mg/kg (rabbit)

SODIUM HYPOCHLORITE (7681-52-9) LD50 (Ingestion): 5800 mg/kg (mouse) TDLo (Ingestion): 1 gm/kg (woman) TDLo (Intravenous): 45 mg/kg (man)

### 12. ECOLOGICAL INFORMATION

Environment WATER: If released to waterways, alkaline products may change the pH of the waterway. Fish will die if the pH

reaches 10-11 (goldfish 10.9, bluegill 10.5) SOIL: May leach to groundwater with toxic effects on aquatic life as above. ATMOSPHERE: May release toxic chlorine gas. Drops or particles released to atmosphere should be removed by gravity and/or be rained out. Hypochlorites are non-persistent in the environment and there is no

accumulation potential as they gradually decompose into a salt and oxygen.

# 13. DISPOSAL CONSIDERATIONS

Waste Disposal For small amounts absorb with sand, vermiculite or similar and dispose of to an approved landfill site. Contact the

manufacturer for additional information if larger amounts are involved. Prevent contamination of drains and waterways as aquatic life may be threatened and environmental damage may result. Neutralise with dilute acid (e.g.

3 mol/L hydrochloric acid).

**Legislation** Dispose of in accordance with relevant local legislation.

## 14. TRANSPORT INFORMATION

### CLASSIFIED AS A DANGEROUS GOODS BY THE CRITERIA OF THE ADG CODE

Shipping Name CORROSIVE LIQUID, N.O.S.

UN No. 1760 DG Class 8 Subsidiary Risk(s) None Allocated

Packing Group III Hazchem Code 2R EPG 8A1

### 15. REGULATORY INFORMATION

Poison Schedule Classified as Schedule 6 (S6) Poison using the criteria in the Standard for the Uniform Scheduling of Drugs

and Poisons (SUSDP).

All chemicals listed on the Australian Inventory of Chemical Substances (AICS).

### **16. OTHER INFORMATION**

Additional Information



#### **ABBREVIATIONS:**

ADB - Air-Dry Basis.

BEI - Biological Exposure Indice(s)

CAS# - Chemical Abstract Service number - used to uniquely identify chemical compounds.

CNS - Central Nervous System.

EINECS - European Inventory of Existing Commercial chemical Substances.

IARC - International Agency for Research on Cancer.

M - moles per litre, a unit of concentration.

mg/m3 - Milligrams per cubic metre. NOS - Not Otherwise Specified.

NTP - National Toxicology Program.

OSHA - Occupational Safety and Health Administration.

pH - relates to hydrogen ion concentration using a scale of 0 (high acidic) to 14 (highly alkaline).

ppm - Parts Per Million.

RTECS - Registry of Toxic Effects of Chemical Substances.

TWA/ES - Time Weighted Average or Exposure Standard.

#### **HEALTH EFFECTS FROM EXPOSURE:**

It should be noted that the effects from exposure to this product will depend on several factors including: frequency and duration of use; quantity used; effectiveness of control measures; protective equipment used and method of application. Given that it is impractical to prepare a Clean Plus Chemicals report which would encompass all possible scenarios, it is anticipated that users will assess the risks and apply control methods where appropriate.

#### PERSONAL PROTECTIVE EQUIPMENT GUIDELINES:

The recommendation for protective equipment contained within this Clean Plus Chemicals report is provided as a guide only. Factors such as method of application, working environment, quantity used, product concentration and the availability of engineering controls should be considered before final selection of personal protective equipment is made.

### **Report Status**

This Safety Data Sheet document has been compiled by Clean Plus Chemicals. Further clarification regarding any aspect of this product should contact Clean Plus Chemicals directly. While Clean Plus Chemicals has taken all due care to include accurate and upto-date information in this SDS, it does not provide any warranty as to accuracy or completeness. As far as lawfully possible, Clean Plus Chemicals accepts no liability for any loss, injury or damage (including consequential loss) which may be suffered or incurred by any person as a consequence of their reliance on the information contained in this SDS.

### **End of Report**

## **Prepared By**

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